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# **Chapter 12**

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The calculations of quantities in  
chemical reactions in a subject of  
chemistry • Mole ratio- a conversion  
factor derived from the coefficients of a

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balanced chemical equation interpreted  
in terms of moles

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## Stoichiometry Vocabulary

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Chapter 12 REVIEW: Stoichiometry,  
Theoretical, Actual & Percent yield Part I.

Stoichiometry 1.  $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$

How many grams of iodine will you  
produce if you begin your rxn with 56.7g  
of zinc iodide? 56.7 g 1 mole 1 mole

253.8 g 319.19 g 1 mole 1 mole = 45.08  
grams of Iodine 2.  $1 \text{ Pb}_2(\text{CO}_3)_3 + 6 \text{ K} \rightarrow$   
 $2 \text{ Pb} + 3 \text{ K}_2\text{CO}_3$

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## Stoichiometry Vocabulary

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#### STOICHIOMETRY 12 Vocabulary Review

Match the correct vocabulary term to each numbered statement. Write the letter Of the correct term on. the line.

Column A

1. the starting materials in a chemical reaction
2. a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms Of moles

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## **Mister Chemistry Welcomes You! - Chemistry teacher at ...**

Chapter 12 Stoichiometry 289 ... Section  
Review Objectives ... Vocabulary •  
stoichiometry Part A Completion Use this  
completion exercise to check your  
understanding of the concepts and  
terms that are introduced in this section.

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Each blank can be completed with a term, short ...

## **05 CTR ch12 7/9/04 3:34 PM Page 289 THE ARITHMETIC OF ...**

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