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May June 2012 Chemistry Paper

Paper 02 – Structured Essays Question 1 Syllabus References: A: 7.1, 7.7, 7.3, 7.4; B2 7.1, 7.2, 7.3 Part (a) (i) tested candidates' knowledge of experimental procedure for collecting and measuring the volume of gas generated during a chemical reaction. Candidates were required to draw a suitable

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A is solution containing 6.22 g of an acid H₂Y per dm³. B contains 3.90 g of NaOH per dm³ of solution. (a) Put A into the burette and titrate it against 20.0 cm³ or 25.0 cm³ portions of B using methyl orange as indicator. Repeat the titration to obtain consistent titres. Tabulate your burette readings and calculate the average volume of acid A used.

Chemistry Paper 1 (Practical) ,May/June 2012

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9701 s12 ms 42

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an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks.

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CIE Paper 6 IGCSE Chemistry Past Papers - PMT

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CIE Paper 3 IGCSE Chemistry Past Papers - PMT

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INTERNATIONAL GCSE CHEMISTRY PAPER 1C - SUMMER 2012 Question number Expected Answer Accept Reject Marks 1 (a) measuring cylinder B Or 2D diagram C Ignore D (top pan) balance /scale(s) / weighing machine A or E Ignore extra apparatus on top of tripod eg beaker Minimum is 2 vertical or diagonal lines with something on top 6

Mark Scheme (Results) Summer 2012 - Edexcel

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This question was attempted by majority of the candidates and the performance was fair. In (a)(i)(ii) and (iii) most candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

Chemistry Paper 2 (Essay) ,May/June 2012

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9701 s12 ms 22

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