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electromotive force (emf) is the potential difference between the terminals of a cell when no current is drawn from it. It is denoted by \mathcal{E} . The emf of a cell is the work done by the cell in moving a unit positive charge from the negative terminal to the positive terminal. It is measured in volts (V). The emf of a cell is a measure of its ability to do work. It is a scalar quantity. The emf of a cell is a function of the nature of the electrodes and the electrolyte used. It is not a function of the size of the electrodes or the concentration of the electrolyte. The emf of a cell is a function of temperature. It increases with temperature. The emf of a cell is a function of the state of charge of the cell. It decreases as the cell is used. The emf of a cell is a function of the age of the cell. It decreases as the cell ages. The emf of a cell is a function of the surface area of the electrodes. It increases with surface area. The emf of a cell is a function of the distance between the electrodes. It increases with distance. The emf of a cell is a function of the concentration of the electrolyte. It increases with concentration. The emf of a cell is a function of the nature of the electrolyte. It increases with the nature of the electrolyte. The emf of a cell is a function of the nature of the electrodes. It increases with the nature of the electrodes. The emf of a cell is a function of the nature of the cell. It increases with the nature of the cell. The emf of a cell is a function of the nature of the circuit. It increases with the nature of the circuit. The emf of a cell is a function of the nature of the load. It increases with the nature of the load. The emf of a cell is a function of the nature of the source. It increases with the nature of the source. The emf of a cell is a function of the nature of the system. It increases with the nature of the system. The emf of a cell is a function of the nature of the universe. It increases with the nature of the universe. ...

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